

1 - PRODUCT AND COMPANY IDENTIFICATION

PRODUCT NAME: SODIUM HYDROXIDE 4% SOLUTION

CHEMICAL NAME/

CLASS/SYNONYMS: None

PRODUCT NUMBER: SODIUM HYDROXIDE 4% SOLUTION

UN/NA NUMBER: 1824 CHEMICAL FAMILY: Sodium salt

CAS NUMBER: Not applicable for mixtures.

FORMULA: NaOH

COMPANY: Lab Alley LLC

22111 Highway 71 West, Suite 601, Spicewood, Texas 78669

Tel.: 512-668-9918 www.laballey.com

EMERGENCY PHONE: InfoTrac: 800-535-5053

DATE PREPARED: February 28, 2019

2 – HAZARDS IDENTIFICATION

GHS HAZARD CLASSIFICATION:

Physical Hazards

Health Hazards

Serious Eye Damage/Irritation: Catagory 1 - Causes severe eye damage

WARNING LABEL ITEMS INCLUDING PRECAUTIONARY STATEMENTS:

Pictograms:







SIGNAL WORD:...... DANGER!

GHS HAZARD AND PRECAUTIONARY STATEMENTS:

H300 H310 H330: Fatal if swallowed, in contact with skin or if inhaled

H341: Suspected of causing genetic defects

H361: Suspected of damaging fertility or the unborn child

P101+102+103: If medical advice is needed, have product container or label at hand. Keep out of the reach of children. Read label before use.

P202+270+280+281: Do not handle until all safety precautions have been read and understood. Do not eat, drink or smoke when using this product. Wear protective gloves/protective clothing/eye protection/face protection. Use personal protective equipment as required.



P501: Dispose of contents/container: Treatment, storage, transportation and disposal must be in accordance with Federal, State/Provincial and Local Regulations. Regulations may vary in different locations. Characterization and compliance with applicable laws are the responsibility solely of the generator. Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

TOTAL VOC's: None

3 - COMPOSITION / INFORMATION ON INGREDIENTS

HAZARDOUS INGREDIENT	US INGREDIENT PERCENT	
Sodium Hydroxide	4	1310-73-2
Water	96	7732-18-5

4 - FIRST-AID MEASURES

BREATHING (INHALATION) :	Remove from exposure area to fresh air immediately. If breathing has
	stopped, perform artificial resuscitation. Keep person warm and at rest.
	Treat symptomatically and supportively. Seek medical attention
	immediately. Qualified medical personnel should consider
	administering oxygen.
SWALLOWING (INGESTION):	Give large amounts of fresh water or milk immediately. Do not give
	anything by mouth if person is unconscious or otherwise unable to
	swallow. If vomiting occurs, keep head below hips to prevent
	aspiration. Treat symptomatically and supportively. Seek medical
	attention immediately.

EYES: Flush eye with copious quantities of water. If persistent irritation occurs, obtain medical attention.

product may have occurred.



5 – FIRE-FIGHTING MEASURES

GENERAL FIRE HAZARDS: Fire fighters should wear full protective clothing, including self-

contained breathing equipment.

AUTOIGNITION TEMP:..... NA

EXTINGUISHING MEDIA: Determined by surrounding material. In case of fire, use water fog, dry

chemical, CO2, or "alcohol" foam.

SPECIAL FIRE FIGHTING

training. Evacuate surrounding areas. Keep unnecessary and unprotected personnel from entering. Do not touch or walk through spilled material. Provide adequate ventilation. Wear appropriate respirator when ventilation is inadequate. Put on appropriate personal

protective equipment. Spilled product may be slippery.

UNUSUAL FIRE AND

EXPLOSION HAZARDS:............. Containers may explode from internal pressure if confined to fire. Cool

with water spray.

6 - ACCIDENTAL RELEASE MEASURES

SPILL PROCEDURES: Wear appropriate personal protective equipment before approaching

spill site. For small spills, dilute with water to sewer if allowed by local and state regulations. If unable to wash product with water, absorb with inert material (sand or other approved material) and dispose of in

accordance with applicable regulations.

WASTE DISPOSAL: Treatment, storage, transportation and disposal must be in accordance

with Federal, State/Provincial and Local Regulations. Regulations may vary in different locations. Characterization and compliance with applicable laws are the responsibility solely of the generator. Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in

accordance with federal, state and local requirements.

7 - HANDLING and STORAGE

STORAGE: Keep in a tightly closed container, stored in a cool, dry, ventilated area

> below 44°C (110°F). Protect against physical damage. Isolate from incompatible substances. Containers of this material may be hazardous when empty since they retain product residues (vapors, liquid); observe all warnings and precautions listed for the product. Drum must not be

washed out or used for other purposes.

HANDLING: Avoid contact with eyes, skin and clothing. Do not inhale vapors and

fumes. Wash thoroughly after handling. Use only with adequate

ventilation. Do not take internally. For industrial use only.



8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

OCCUPATIONAL EXPOSURE LIMITS

HAZARDOUS INGREDIENT PEL TLV-TWA
Sodium Hydroxide 2 mg/m³ 2 mg/m³

Water None Established None Established









EXPOSURE CONTROLS:

Good general ventilation (typically 10 air changes per hour) should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.

RESPIRATORY PROTECTION: If engineering controls do not maintain airborne concentrations below

recommended exposure limits (where applicable) or to an acceptable level (in countries where exposure limits have not been established), an approved respirator must be worn. In the United States of America, if respirators are used, a program should be instituted to assure compliance with OSHA Standard 63 FR 1152, January 8, 1998. Respirator type: Air-purifying respirator with an appropriate, government approved (where applicable), air-purifying filter, cartridge or canister. Contact health and safety professional or manufacturer for specific information.

PROTECTIVE CLOTHING:

Eye/face protection: Wear chemical goggles; face shield (if splashing is possible). **Skin protection:** Chemical resistant, impermeable gloves. Gloves should be tested to determine suitability for prolonged contact. Use of impervious apron or chemical suit and chemical resistant boots are recommended.

ADDITIONAL MEASURES:

Handle in accordance with good industrial hygiene and safety practice. Wash thoroughly with soap and water after handling and before eating, drinking, or using tobacco. Safety shower and eye wash should be available close to work areas.

9 - PHYSICAL / CHEMICAL PROPERITES

BOILING POINT: $212^{\circ}\text{F} (100^{\circ}\text{C})$ **FREEZING POINT:** $35^{\circ}\text{F} (1.66^{\circ}\text{C})$

FLASHPOINT:......Non-flammable material

UPPER FLAME LIMIT (%): NA LOWER FLAME LIMIT (%): ... NA VAPOR PRESSURE: ND



VAPOR DENSITY (AIR=1):..... > 1

SPECIFIC GRAVITY: 1.035 - 1.045

SOLUBILITY IN WATER:..... Complete

VOLATILITY

INCLUDING WATER: 8.69 pounds per gallon

10 - STABILITY and REACTIVITY

STABILITY: Stable

HAZARDOUS DECOMP.:..... Will not occur

INCOMPATIBILITY: Avoid direct contact with strong acids. Add slowly to water or acids

with dilution and agitation to minimize the possible exothermic reaction. Avoid contact with aluminum, tin, zinc, leather, and organic halogen or nitro compounds. Sodium hydroxide in contact with acids and organic halogen compounds, especially trichloroethylene, may cause violent reactions. Contact with nitromethane and other similar nitro compounds causes formation of shock-sensitive salts. Contact with metals such as aluminum, magnesium, tin, and zinc cause formation of flammable hydrogen gas. Sodium hydroxide, even in fairly dilute solution, reacts readily with various sugars to produce carbon

monoxide. Precautions should be taken including monitoring the tank atmosphere for carbon monoxide to ensure safety of personnel before

vessel entry.

HAZARDOUS REACTIONS: Not expected to be Explosive, Self-Reactive, Self-Heating, or an

Organic Peroxide under US GHS Definition(s).

11 - TOXICOLOGICAL INFORMATION

THRESHOLD LIMIT VALUE: 2 mg/m³
OSHA PEL: 2 mg/m³

LISTED CARCINOGEN: This product IS NOT listed in the National Toxicology Program (NTP)

Report on Carcinogens (latest edition) or has been found to be a potential carcinogen in the International Agency for Research on Cancer (IARC) Monographs (latest editions) or found to be a potential

carcinogen by OSHA.

MEDICAL CONDITION

AGGRAVATED: Pre-existing medical conditions of the following organ(s) or organ

system(s) may be aggravated by exposure to this material: Respiratory

system. Eyes. Skin.



such exposures.

Safety Data Sheet (SODIUM HYDROXIDE 4% SOLUTION)

INFORMATION ON ACUTE TOXICOLOGICAL EFFECTS

ORAL				
Product:				
a burning sensation in the mouth, corrosion of thelips, mouth, tongue and pharynx, and severe esophageal				
and abdominal pain, vomiting of blood and large pieces of mucosa, and bloody diarrhea. Asphyxia can				
occur from swelling of the throat. Mediastinitis, alkalemia, pallor, weak, slow pulse, cardiovascular				
collapse, shock, coma and death may occur. Perforation of the alimentary tract and constrictive scarring				
may result. Esophageal stricture may occur weeks, months, or even years later to make swallowing difficult.				
The estimated fatal dose in man is 5grams. Cases of squamous cell carcinoma of the esophagus have				
occurred with latent periods of 12 to 42 years after ingestion. These cancers were believed to be sequela of				
tissue destruction and possibly scar formation rather than the result of direct carcinogenic action of sodium				
hydroxide.				
DERMAL				
Product:				
skin fissures and white eschars may occur without immediate pain. Exposure to solutions as weak as 0.03 N				
(0.12%) for 1 hour has caused injury to healthy skin. With solutions of 0.4-4%, irritation does not occur				
until after several hours. Solutions of 25-50% caused no sensation of irritation within 3 minutes in human				
subjects. Skin biopsies from human subjects having 1 N sodium hydroxide applied to their arms for 15 to				
180 minutes showed progressive changes beginning with dissolution of the cells in the horny layer and				
progressing through edema to total destruction of the epidermis in 60 minutes. A 5% aqueous solution				
caused severe necrosis to the skin of rabbits when applied for 4 hours. Alkalies penetrate the skin slowly.				
The extent of injury depends on the duration of contact. If sodium hydroxide is not removed from the skin,				
severe burn swith deep ulceration may occur. Exposure to the dust or mist may cause multiple small burns				
and temporary loss of hair. Pathologic findings due to alkalies may include gelatinous, necrotic areas at the				
site of contact.				
INHALATION				
Product: Effects due to inhalation of dusts or mist may vary from mild irritation of the nose at 2				
mg/m ³ to severe pneumonitis depending on the severity of exposure. Low concentrations may cause mucous				
membrane irritation with sore throat, coughing, and dyspnea. Intense exposures may result in destruction of				
mucous membranes and delayed pulmonary edema or pneumonitis. Shock may occur.				
REPEATED DOSE TOXICITY				
Product: Anepidemiologic study of 291 workers chronically exposed to caustic				
dusts for 30 years or more found no significant increase in mortality in relation to duration or intensity of				
such exposures.				
SKIN CORROSION / IRRITATION				
Product: Effects are dependent upon concentration and duration of exposure.				
Dermatitis or effects similar to those for acute exposure may occur.				
SERIOUS EYE DAMAGE / IRRITATION				
Product:				
corneal epithelium, corneal opacification, marked edemaand ulceration. After 7 to 13 days either gradual				
recovery begins or thereis progression of ulceration and corneal opacification. Complications of severe eye				
burns are symblepharon with overgrowth of the cornea by a vascularized membrane, progressive or				
recurrent corneal ulceration and permanent corneal opacification. Blindness may occur.				
RESPIRATORY OR SKIN SENSITIZATION				
Product:				

dusts for 30 years or more found no significant increase in mortality in relation to duration or intensity of



MUTAGENCITY

TV LYTTE O		
IN VITRO Product:No Data A	voilable	
IN VIVO	Available	
Product: No Data A	vailable	
Specified Substance(s)	Information as provided by manufacturer	
Sodium Hydroxide 4% Solution	No Data Available	
CARCINOGENICITY		
	nufacturer: After extensive testing and more than 100 years of	
absorbed through the skin. It is not a known ca	s no evidence that caustic is a skin sensitizer or is readily urcinogen, mutagen, developmental toxicant or reproductive	
toxicant. REPODUCTIVE TOXICITY		
	available data the classification criteria are not met. Not	
classified as hazardous.		
SPECIFIC TARGET ORGAN TOXICITY	– SINGLE EXPOSURE	
Product: INHALATION: Exposure	to vapor, mist or liquid can produce burns of the respiratory	
tract. Severe exposures could result in chemic	al pneumonia. EYES: Contact can cause severe damage	
including burns and blindness. The severity of	the effects depend on concentration and how soon after	
	ve. Contact may cause burns and tissue destruction. Note	
	lay between the time the exposure occurs and when the sense	
· · · · · · · · · · · · · · · · · · ·	as much as hours for a dilute solution (0.04%) to minutes	
	Prolonged or repeated contact, even to dilute concentrations,	
0 0	INGESTION: Corrosive. Severe burns and complete tissue	
perforation of mucous membranes of mouth, the		
SPECIFIC TARGET ORGAN TOXICITY		
	CTS: Not available. MUTAGENIC EFFECTS: Not	
	available. DEVELOPMENTAL TOXICITY: Not	
available. The substance is toxic to lungs. Repeated or prolonged exposure to the substance can produce		
	contact with spray mist may produce chronic eye irritation	
	ed exposure to spray mist may produce respiratory tract	
	ial infection. Repeated exposure to a highly toxic material	
	an accumulation in one or many human organs.	
ASPIRATION HAZARD		
	f the product aspirated into the lungs through ingestion or	
vomiting may cause chemical pneumonia.		
OTHER ADVERSE EFFECTS		
Product: No data av	/ailable	

12 – ECOLOGICAL INFORMATION

ACUTE TOXICITY

FISH	
Product:	Bluegill sunfish: 48-hour LC50 = 99 mg/L Mosquito fish: 96-hour
LC50 = 125 mg/L Brown shrimp ((Crangon crangon): 48-hour LC50 = 30 - 100 mg/L
AQUATIC INVERTEBRATES	
Product:	Daphnia magna - Water flea: Acute LC50 196 mg/L Marine water.



CHRONIC TOXICITY FISH **Product:** Concentrations of 20 to 100 mg/L have been reported to kill salmon, trout, carp and crayfish. AQUATIC INVERTEBRATES Product: Expected to have low toxicity: 10 < LC/EC/IC50 <= 100 mg/l TOXICITY TO AQUATIC PLANTS **Product:.....** Freshwater algae are destroyed above pH 8.5. PERSISTENCE AND DEGRADABILITY BIODEGRADATION applicable to predominately inorganic substances. BIOLOGICAL OXYGEN DEMAND **Product:** No data available CHEMICAL OXYGEN DEMAND **Product:.....** No data available BOD / COD RATIO Product: No data available BIOACCUMULATIVE POTENTIAL **Product:** Sodium hydroxide does not bioaccumulate due to its high solubility in water. It is considered slightly toxic to aquatic organisms unless there is a significant pH shift outside the range of 5 - 10; this change may be toxic to aquatic organisms. MOBILITY IN SOIL **Product:** Expected to partition to water. The pH effect of sodium hydroxide in water is naturally reduced by the absorption of atmospheric carbon dioxide. This reduction is also effected by dilution with water and by the natural acidity of a given water body. There is no degradation of sodium hydroxide in waters, only loss by absorption or through chemical neutralization. RESULTS OF PBT AND mPvB ASSESSMENT fulfilling vPvB (very persistent, very bioaccumulative) criteria. OTHER ADVERSE EFFECTS photochemical ozone creation potential, endocrine disruption, global warming potential) are expected from

13 - DISPOSAL CONSIDERATIONS

this product.

WASTE DISPOSAL: Treatment, storage, transportation and disposal must be in accordance with Federal, State/Provincial and Local Regulations. Regulations may vary in different locations. Characterization and compliance with applicable laws are the responsibility solely of the generator. Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.



14 - TRANSPORTATION INFORMATION

Important Note: Shipping descriptions may vary based on mode of transport, quantities, package size, and/or origin and destination. Consult your company's Hazardous Materials/Dangerous Goods expert for information specific to your situation.



UN/NA NUMBER: 1824

PROPER SHIPPING NAME: Sodium Hydroxide, solution

HAZARD CLASS:.....8 PACKAGING GROUP :......

LETTER:..... C (Corrosive substances)

ENVIRONMENTAL HAZARD: Environmental Hazard Value Score (IRCH) = 29. Caustic soda does not

bioaccumulate due to its high solubility in water. It is considered slightly toxic to aquatic organisms unless there is a significant pH shift outside the range of 5-10, which may be toxic to aquatic organisms.

REPORTABLE QUANTITY: 17,120 pounds based on Sodium Hydroxide in mixture.

15 - REGULATIONS

This Safety Data Sheet conforms to ANSI Z400.5, and to the format requirements and the International Chemical Safety Cards of the Global Harmonizing System. This SDS complies with 29 CFR 1910.1200 (HAZARD COMMUNICATION STANDARD). **IMPORTANT:** Read this SDS before handling & disposing of this product. Pass this information on to employees, customers, & users of this product.

EPA SRA Title III Chemical Listings:

TSCA STATUS:..... This product is listed on the TSCA inventory. If this product is a blend,

all ingredients in the product are listed on the TSCA Inventory List.

Any impurities present in this product are exempt from listing.

SARA SECTION 313: SECTION 313 REPORTABLE INGREDIENTS (40 CFR 372):

This product contains toxic chemicals subject to the reporting requirements of Section 313, Title III of the SARA (Superfund Amendments and Reauthorization Act) of 1986: Sodium Hydroxide

(Caustic Soda)

 ACUTE:
 Yes

 CHRONIC:
 Yes

 FIRE:
 No

 PRESSURE:
 No

 REACTIVE:
 Yes

 CLEAN WATER ACT:
 None

IMDG - International Marine Dangerous Goods Code

UN1824, Sodium Hydroxide, Solution, 8, PG II. EmS F-A, S-B. Marine Pollutant: No.

IATA

UN1824, Sodium Hydroxide, Solution, 8, PG II

DEA Chemical Trafficking Act:.. No



16 - OTHER INFORMATION

HMIS*		
HEALTH	3	
FLAMMABILITY	0	
REACTIVITY	0	
PERSONAL PROTECTI	ON H	

*HMIS®HAZARD INDEX: 0=Minimal Hazard, 1=Slight Hazard, 2=Moderate Hazard, 3=Serious Hazard, 4=Severe Hazard. HMIS® rating involves data interpretations that may vary from company to company. They are intended only for rapid, general identification of the magnitude of the specific hazard. To deal adequately with the safe handling of this material, all the information contained in this SDS and product label must be considered.

ND = No Data, NA = Not Applicable/Not Available, $\leq = Less than or equal to$, $\geq = Greater than or equal to$

REVISION STATEMENT: Changes have been made throughout this Safety Data Sheet (SDS). Please read the entire document. Prepared according to the OSHA Hazard Communication Standard (29 CFR 1910.1200) and The Globally Harmonized System of Classification and Labeling of Chemicals (GHS) by the Company Health and Risk Assessment Unit.

DISCLAIMER:

Although the information and recommendations set forth herein (hereinafter "Information") are presented in good faith and believed to be correct as of the date hereof, the Company makes no representations as to the completeness or accuracy thereof. Information is supplied upon the condition that the persons receiving this Safety Data Sheet (SDS) will make their own determination as to its suitability for their intended purposes prior to use. Since the product is within the exclusive control of the user, it is the user's obligation to determine the conditions of safe use of this product. Such conditions should comply with all Federal and State Regulations concerning the Product. It must be recognized that the physical and chemical properties of any product may not be fully understood and that new, possibly hazardous products may arise from reactions between chemicals. The information given in this data sheet is based on our present knowledge and shall not constitute a guarantee for any specific product features and shall not establish a legally valid contractual relationship. REPRESENTATIONS OR WARRANTIES, **EITHER EXPRESS** OR IMPLIED, MERCHANTABILITY, FITNESS FOR A PARTICULAR PURPOSE OR OF ANY OTHER NATURE ARE MADE HEREUNDER WITH RESPECT TO INFORMATION OR THE PRODUCT TO WHICH INFORMATION REFERS.

This is the last page of this SDS