



SAFETY DATA SHEET

SECTION 1: Identification of the substance/mixture and of the company/undertaking

1.1 Product identifiers

Product name Nitric Acid 1M Solution, Lab Grade

CAS number 7697-37-2

Synonyms Hydrogen nitrate, Aqua fortis, Azotic acid, Salpetersaeure,

Nitrooxidanyl

1.2 Relevant identified uses of the substance or mixture and uses advised against

Identified uses Laboratory chemicals.

1.3 Details of the supplier of the safety data sheet

Company Lab Alley, LLC

12501 Pauls Valley Road Austin, Texas 78737

U.S.A.

Telephone 512-668-9918 Fax 512-886-4008

1.4 Emergency telephone

Emergency Phone # US & Canada: 1-800-535-5053 INFOTRAC

International 1-352-323-3500 INFOTRAC

SECTION 2: Hazards identification

2.1 Classification of the substance or mixture

GHS Classification in accordance with 29 CFR 1910 (OSHA HCS)

Oxidizing liquidsCategory 3Serious eye damageCategory 1Skin corrosionCategory 1B

2.2 GHS Label elements, including precautionary statements

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Signal Word Danger

Hazard statements May intensify fire; oxidizer. Causes severe skin burns and eye

damage.

Precautionary statements

If medical advice is needed, have product container or label at hand.

Keep out of reach of children.

Read label before use.

Keep away from heat/sparks/open flames/hot surfaces.

No smoking.

Wear protective gloves/protective clothing/eye protection/face protection.

Do not breathe dust/fume/gas/mist/vapors/spray.

Do not eat, drink, or smoke when using this product.

Take any precaution to avoid mixing with combustibles.

Keep and store away from clothing/combustible materials.

Wash skin thoroughly after handling.

IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do. Continue rinsing. Immediately call a POISON CENTER or doctor/physician.

IF ON SKIN/HAIR: Remove/Take off all contaminated clothing immediately. Rinse skin with water/shower. Wash contaminated clothing before reuse.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

In case of fire: Use reagents recommended in Section 5 for extinction.

Store locked up.

Dispose of contents/container according to local regulations.

2.3 Hazards not otherwise classified (HNOC) or not covered by GHS

None identified.

SECTION 3: Composition/information on ingredients

3.1 Components

	Chemical name	Chemical name Common name and synonyms		Concentration
Ī	Nitric Acid	Hydrogen nitrate, Aqua fortis, Azotic acid	7697-37-2	9.217 %
	Deionized Water	Aqua, Deionized Water, Dihydrogen Oxide	7732-18-5	90.78%

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SECTION 4: First aid measures

4.1 Description of first-aid measures

General advice

If inhaled Move exposed individual to fresh air. Loosen clothing as necessary and

position individual in a comfortable position. Seek medical advice if

discomfort or irritation persists.

In case of skin contact Wash affected area with soap and water. Rinse or flush skin/hair gently with

water for at least 30 minutes. Seek immediate medical attention.

In case of eye contact Protect unexposed eye. Remove contact lens(es) if able to do so during

rinsing. Rinse or flush eye gently with water for at least 30 minutes, lifting upper and lower lids. Seek immediate medical attention (opthalmologist).

If swallowed Rinse mouth thoroughly. Do not induce vomiting. Have exposed individual

drink sips of water. Seek medical attention if irritation, discomfort, or

vomiting persists.

4.2 Most important symptoms and effects, both acute and delayed

Headache; shortness of breath; irritation/burns, all routes of exposure. May cause severe burns, blindness, and/or permanent damage. May cause burns, deep penetrating ulcerations of the skin, delayed tissue destruction, redness, and pain. May cause gastrointestinal irritation with nausea, vomiting and diarrhea.

4.3 Indication of any immediate medical attention and special treatment needed

If seeking medical attention, provide SDS document to physician.

SECTION 5: Firefighting measures

5.1 Extinguishing media

Suitable extinguishing media Does not burn. Use extinguishing media

appropriate for surrounding fire. If in laboratory setting, follow laboratory fire suppression procedures. Use appropriate fire suppression agents for adjacent combustible materials or

sources of ignition.

Unsuitable extinguishing mediaNo information available.

5.2 Specific hazards arising from the substance or mixture

Combustion products may include Carbon oxides, Nitrogen oxides (NOx), or other toxic vapors.

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5.3 Special protective equipment and precautions for firefighters

Move product containers away from fire or keep cool with water spray as a protective measure, where feasible.

5.4 Further information

Flash Point No information available.

Autoignition Temperature No information available.

Explosion limits

Upper No data available.Lower No data available.

Sensitivity to Mechanical Impact No information available.

Sensitivity to Static Discharge No information available.

NFPA

Health	Flammability	Instability	Physical hazards
3	0	0	OX

SECTION 6: Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Wear protective equipment. Use respiratory protective device against the effects of fumes/dust/aerosol. Keep unprotected persons away. Ensure adequate ventilation. Keep away from ignition sources. Protect from heat. Stop the spill, if possible. Contain spilled material by diking or using inert absorbent. Transfer to disposal or recovery container.

6.2 Environmental precautions

Prevent from reaching drains, sewer, or waterway.

6.3 Methods and materials for containment and cleaning up

If in a laboratory setting, follow Chemical Hygiene Plan procedures. Collect liquids using vacum or by use of absorbents. Place into properly labeled containers for recovery or disposal. If necessary, use trained response staff/contractor.

6.4 Reference to other sections

See section 2 for full list of hazard and precaution statements.

SECTION 7: Handling and storage

7.1 Precautions for safe handling

Precautions on safe handling

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Prevent formation of aerosols. Follow good hygiene procedures when handling chemical materials. Do not eat, drink, smoke, or use personal products when handling chemical substances. If in a laboratory setting, follow Chemical Hygiene Plan. Use only in well-ventilated areas. Avoid splashes or spray in enclosed areas. Keep away from heat and sources of ignition.

Hygiene measures

Handle in accordance with good industrial hygiene and safety practice.

7.2 Conditions for safe storage, including any incompatibilities

Storage conditions

Store in a cool location. Provide ventilation for containers. Avoid storage near extreme heat, ignition sources, or open flame. Store away from foodstuffs and oxidizing agents. Store in cool, dry conditions in well-sealed containers. Keep container tightly sealed. Store with similar hazards.

Incompatibilities

Highly reactive with alkalis, reducing agents, combustible materials, organic materials, metals, acids, and aldehydes.

SECTION 8: Exposure controls/personal protection

8.1 Occupational exposure limits

US. OSHA Table Z-1 Limits for Air Contaminants (29 CFR 1910.1000)

Component	Туре	Value
Nitric acid	TWA	2 ppm

US. ACGIH Threshold Limit Values

Component	Type	Value
Nitric acid	TWA	2 ppm

US. NIOSH: Pocket Guide to Chemical Hazards

Component	Туре	Value		
Nitric acid	TWA	2 ppm		

Biological occupational exposure limits

No information available.

8.2 Exposure controls

Appropriate engineering controls

Emergency eye wash fountains and safety showers should be available in the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or mists below the applicable workplace exposure limits indicated above.

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Personal protective equipment

Eye/face protection

Safety glasses with side shields or goggles.

Skin protection

The glove material has to be impermeable and resistant to the product being handled. Selection of glove material on consideration of penetration times, rates of diffusion, and degradation.

Body Protection

Wear appropriate protective gloves and clothing to prevent skin exposure.

Respiratory protection

Not required under normal conditions of use. Use suitable respiratory protective device when high concentrations are present. Use suitable respiratory protective device when aerosol or mist is formed. For spills, respiratory protection may be advisable.

Control of environmental exposure

No information available.

SECTION 9: Physical and chemical properties

9.1 Information on basic physical and chemical properties

Physical State Liquid

Appearance Colorless to a pale yellow

Odor Strong acrid

Odor Threshold No information available

pH < 1.0

Melting Point/Range No information available
Boiling Point/Range No information available
Evaporation Rate No information available
Flammability (solid) No information available
Flammability or explosive limit No information available

Upper

Lower

Vapor PressureNo information availableVapor DensityNo information availableDensityNo information available

Soluble in water

Partition coefficient; No information available

n-octanol/water

Autoignition Temp
Decomposition Temp
Viscosity
No information available
No information available
No information available

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Molecular Formula HNO3

Molecular Weight 63.013 g/mol

VOC Content(%)

Oxidizing properties

No information available

No information available

9.2 Other safety information

No information available.

SECTION 10: Stability and reactivity

10.1 Reactivity

Oxidizer. Reacts violently with alcohol, organic material, turpene, charcoal. Violent reaction with Nitric acid, Acetone, and Sulfuric acid. Nitric acid will react with water or steam to produce heat and toxic, corrosive, flammable vapors.

10.2 Chemical stability

No decomposition if used and stored according to specifications.

10.3 Possibility of hazardous reactions

Oxidizer: contact with combustible/organic material may cause fire.

10.4 Conditions to avoid

Excess heat, combustible materials, and incompatible materials.

10.5 Incompatible materials

Highly reactive with alkalis, reducing agents, combustible materials, organic materials, metals, acids, and aldehydes.

10.6 Hazardous decomposition products

Nitrogen oxides (NOx).

SECTION 11: Toxicological information

11.1 Information on toxicological effects

Product Information, Component Information

Acute toxicity

Component LD50 Oral		LD50 Dermal	LC50 Inhalation	
Nitric acid	430 mg/kg (Rat)	-	67 ppm 4 h (Rat)	

Skin corrosion/irritation

Causes severe skin burns.

Serious eye damage/eye irritation

Corrosive to eyes; causes serious eye damage.

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Respiratory or skin sensitization

No additional information.

Germ cell mutagenicity

No additional information.

Carcinogenicity

Component	CAS	IARC	NTP	ACGIH	OSHA	Mexico
Nitric acid	7697-37-2	Not listed				

Specific target organ toxicity - single exposure

No additional information.

Specific target organ toxicity - repeated exposure

No additional information.

Reproductive toxicity

Experiments have shown reproductive toxicity effects on laboratory animals.

Chronic effects

No additional information.

11.2 Additional Information

No information available.

SECTION 12: Ecological information

12.1 Toxicity

No information available.

12.2 Persistence and degradability

Readily degradable in the environment.

12.3 Bio accumulative potential

No information available.

12.4 Mobility in soil

Aqueous solution has high mobility in soil.

12.5 Results of PBT and vPvB assessment

No information available.

12.6 Endocrine disrupting properties

No information available.

12.7 Other adverse effects

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SECTION 13: Disposal considerations

13.1 Waste Disposal Methods

Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. Chemical waste generators must also consult local, regional, and national hazardous waste regulations to ensure complete and accurate classification.

SECTION 14: Transport information

DOT (US)

UN-No UN2031
Proper Shipping Name Nitric acid

Hazard Class 8
Packing Group ||

IMDG

UN-No UN2031
Proper Shipping Name Nitric acid

Hazard Class 8
Packing Group ||

IATA

UN-No UN2031
Proper Shipping Name Nitric acid

Hazard Class 8
Packing Group ||

SECTION 15: Regulatory information

US federal regulations This product is a "Hazardous Chemical" as defined by the OSHA Hazard Communication Standard, 29 CFR 1910.1200.

TSCA Section 12(b) Export Notification (40 CFR 707, Subpt. D)

Not applicable.

CERCLA Hazardous Substance List (40 CFR 302.4)

7697-32-2 Nitric acid 1000 lbs.

SARA 304 Emergency release notification

7697-32-2 Nitric acid 1000 lb.

OSHA Specifically Regulated Substances (29 CFR 1910.1001-1053)

Not regulated.

Superfund Amendments and Reauthorization Act of 1986 (SARA)

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SARA 302 Extremely hazardous substance

7697-32-2 Nitric acid 1000 lb.

SARA 311/312 Hazardous

Acute Health Hazard.

SARA 313 (TRI reporting)

Listed, 7697-32-2 Nitric Acid.

Other federal regulations

Clean Air Act (CAA) Section 112 Hazardous Air Pollutants (HAPs) List

Not regulated.

Clean Air Act (CAA) Section 112(r) Accidental Release Prevention (40 CFR 68.130)

Not regulated.

Safe Drinking Water Act

Not regulated.

FEMA Priority Substances Respiratory Health and Safety in the Flavor Manufacturing Workplace

Not listed.

US state regulations

US. Massachusetts RTK - Substance List

Listed.

US. New Jersey Worker and Community Right-to-Know Act

Listed.

US. Pennsylvania Worker and Community Right-to-Know Law

Listed.

California Proposition 65

Not listed.

SECTION 16: Other information

Issue date: 09/06/2024

SECTION 17: Disclaimer

The information provided on this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date of its publication. The information given is designed only as a guide for safe handling, use, processing, storage, transportation, disposal and release and is not to be considered as a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other material or in any process, unless specified in the text.

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